

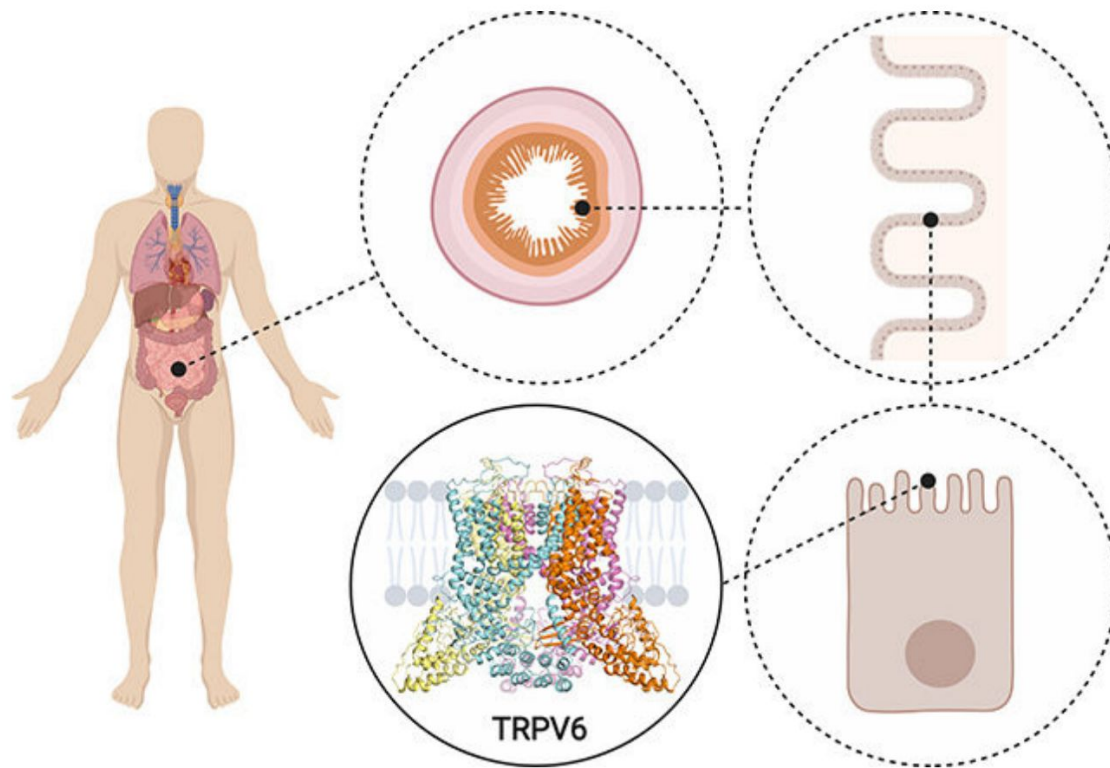


MAGNESIUM BINDING TO TRPV6 ION CHANNEL: INSIGHTS FROM MOLECULAR MODELING

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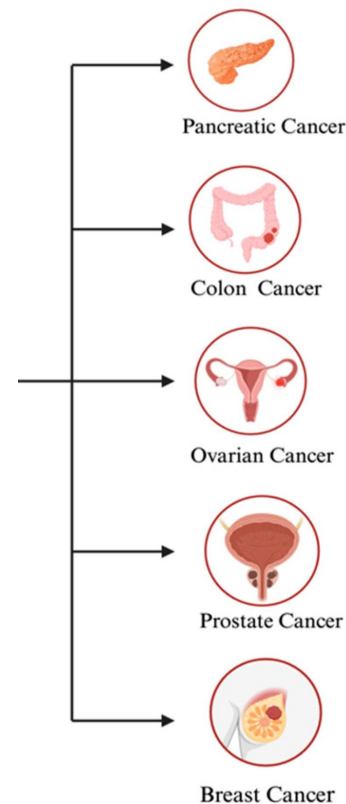
BCADD, 21st October 2025

TRPV6 ion channel



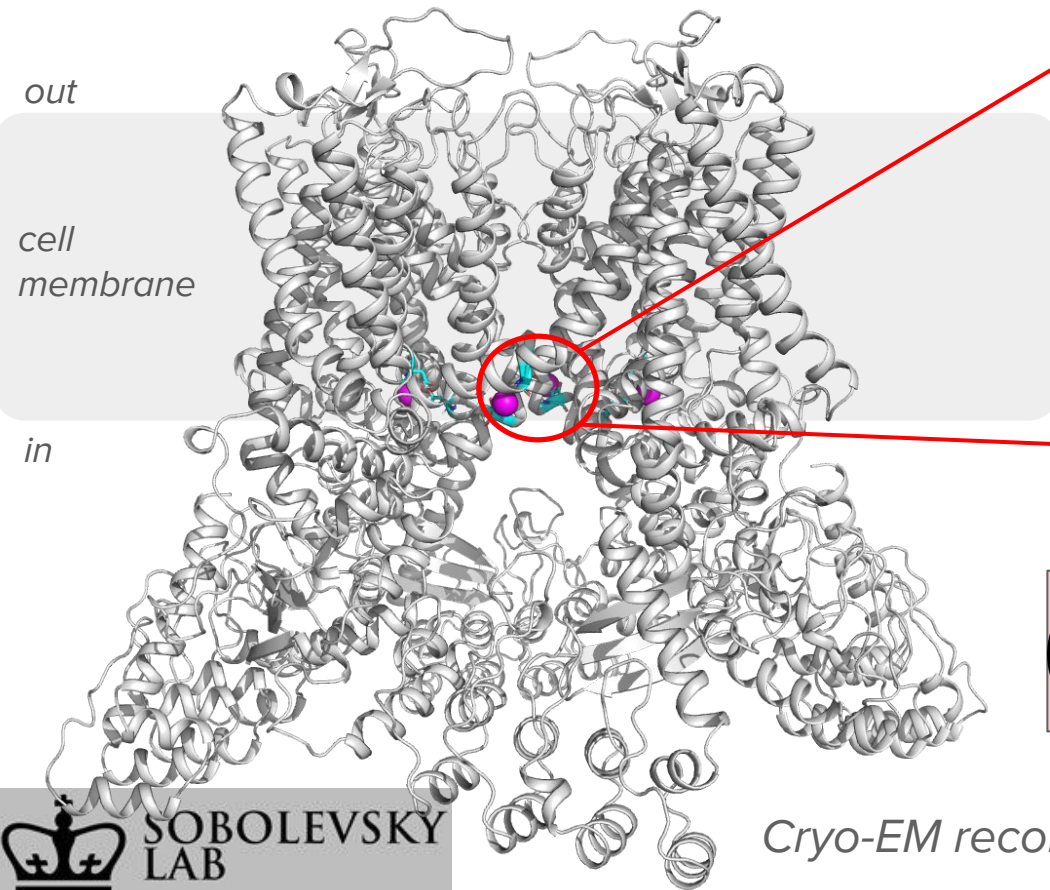
(Yelshanskaya et al, 2021)

TRPV6 overexpression

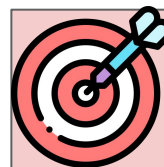
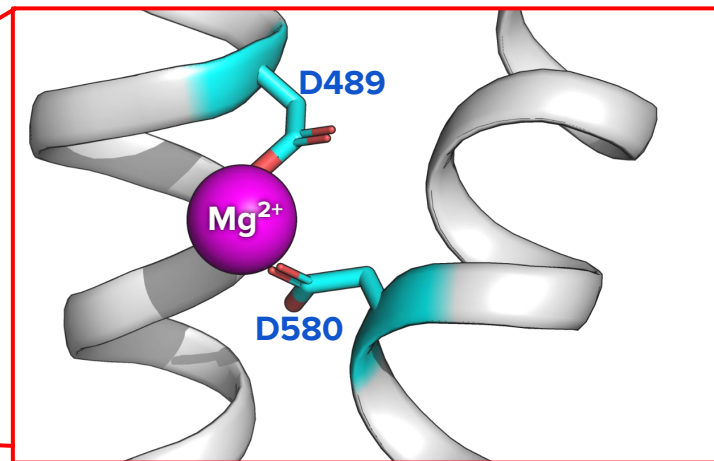


(Wang et al, 2024)

Problem statement



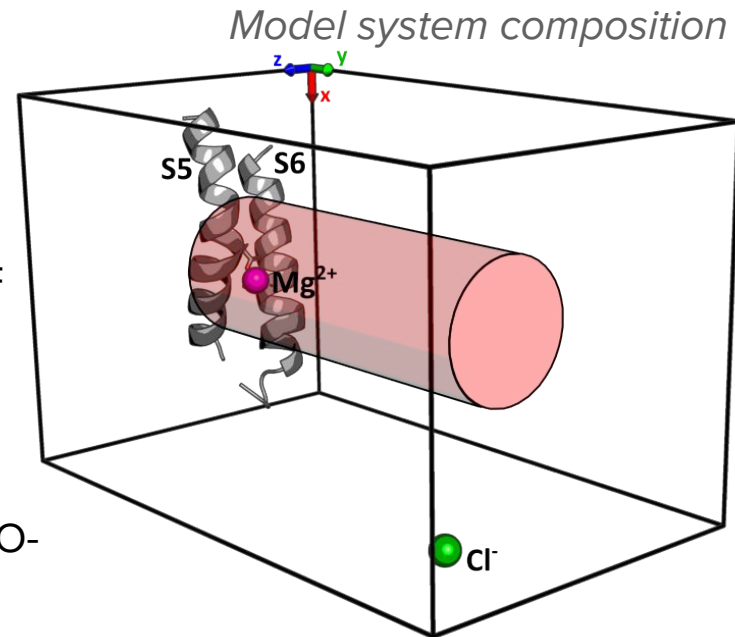
potential Mg^{2+} ion binding site



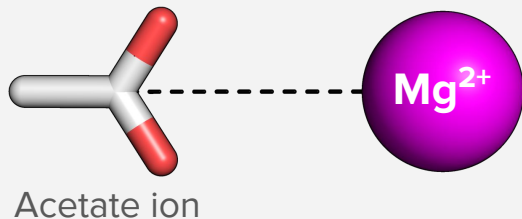
prove Mg^{2+} selectivity of
D489-D580 site

Methodology

- Model system: **D489-D580 site** with **Mg²⁺** or **Ca²⁺**
- **Electronic Continuum Correction** + Amber-99sd-ildn FF
- **Well-Tempered Metadynamics** with CVs:
 - **CN_w** - number of H₂O oxygen atoms in the 1st coordination shell of the cation;
 - **L₁, L₂** - distances between the carbon atoms of COO- groups and the cation.



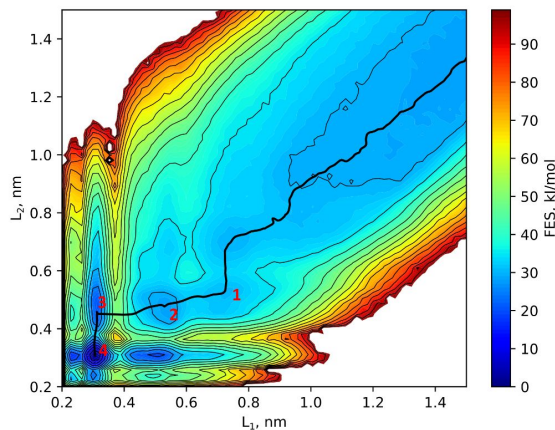
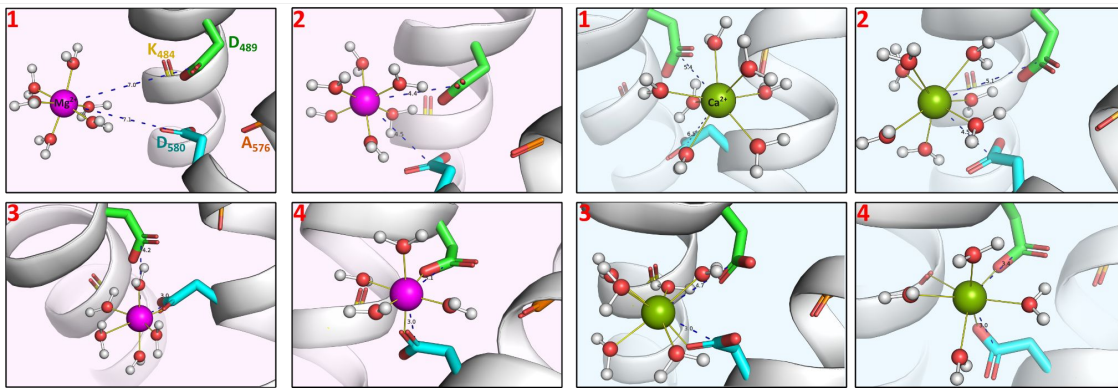
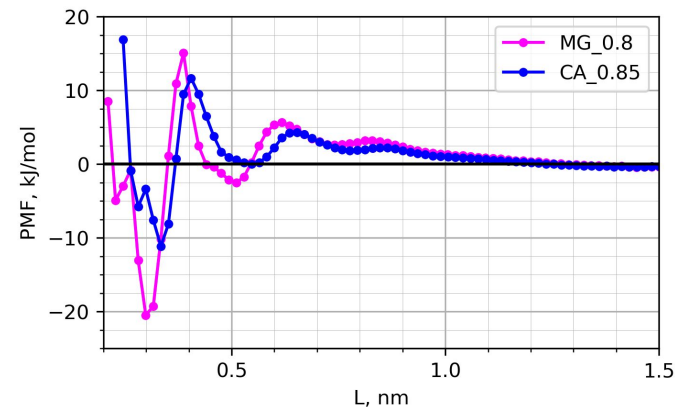
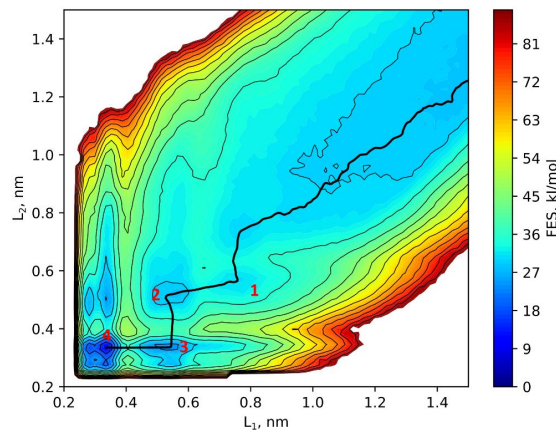
Verification



ΔG^0 (kJ/mol) CH₃COO⁻-cation system with different ECC scaling factors (SF)

	Experiment	SF=0.8	SF=0.85	SF=1
Mg ²⁺	(-7.3) - (-3.4)	-3.7 ± 0.1	-	6 ± 13
Ca ²⁺	(-6.8) - (-2.5)	0.0 ± 0.4	-3.0 ± 0.2	-11.7 ± 0.1

Modeling of Mg^{2+} / Ca^{2+} binding to D489-D580

 Mg^{2+}  Ca^{2+} 

$$\Delta G^0(\text{Mg}^{2+}) = -12.9 \pm 0.4 \text{ kJ/mol}$$

$$\Delta G^0(\text{Ca}^{2+}) = -3.7 \pm 0.6 \text{ kJ/mol}$$

~35 times preferability of Mg^{2+}

Conclusion

The **developed protocol** for calculating binding free energy (ΔG^0) at ion-binding sites was **validated using experimental** CH_3COO^- data.

A **novel** intracellular site (**D489-D580**) in TRPV6 shows **high selectivity for Mg^{2+}**

These results **aligns** with the findings from **cryo-electron microscopy** and **electrophysiology** studies

Proposed mechanism:
intracellular Mg^{2+} binding to **D489-D580** site stabilizes TRPV6 in a closed state

Significance:

The obtained results open new avenues for developing therapies targeting TRPV6-associated diseases, including cancers.



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Thank you for attention!

Neuberger, A., Shalygin, A., Veretenenko, I. et al.
**The locking mechanism of human TRPV6
inhibition by intracellular magnesium.**
Nat Commun (in press)

